



ANRA's Water Quality Monitoring Program: What We Test for and Why

July 11, 2017





Sampling Considerations

ICEC Surface Water Quality Monitoring Procedures, Volume 1: Physical and Chemical Monitoring Methods Google: "TCEQ RG-415" Water Quality Planning Division printed on recycled pape TEXAS COMMISSION ON ENVIRONMENTAL QUALITY

RG-415 Revised August 2012

- Frequency and timing of sampling
- Type of sampling
- Where to sample
- Order of sample collection
- What parameters to collect



What Parameters to Collect

- Depends on purpose of sampling and watershed factors
 - Trend analysis
 - Watershed protection/drinking water supply protection
 - Permit support monitoring
- Basic Parameters
- Additional Parameters as needed

Parameter	Parameter Code	Final Form for Field Data (Rounding, Significant Figures)			
Water temperature (°C)	00010	Report temperature to the nearest tenth in degrees Celcius. (example: 25.94 to 25.9, or 26.97 to 27.0)			
pH (s.u.)	00400	Report pH to the nearest tenth in pH standard units. (example: 7.94 to 7.9, or 7.97 to 8.0)			
DO (mg/L)	00300	Report dissolved oxygen to the nearest tenth in mg/L. (example: 5.94 to 5.9, or 6.97 to 7.0)			
Specific conductance (µS/cm)	00094	Report specific conductance to only three significant figures if the value exceeds 100—for example: 1532 to 1530. Do not report ORP, which is displayed by some multiprobe instruments. For values < 100 μ S/cm follow standard rounding rules and report the nearest whole number. For example, report 88.7 as 88 μ S/cm.			
Salinity (ppth)	00480	Report salinity values above 2.0 ppth to the nearest tenth in parts per thousand. Do not report salinity from freshwater or inland (brine) locati In estuarine waters report the actual values displayed by the instrument above 2.0 ppt, and values less than 2.0 as < 2.0 (examples: 0.85 to < 2.0 (2.5 to < 2.0) Determine if a station is estuarine (experiencing periods salinity is > 2.0 ppth), and always report salinity at this station, regardle salinity during periods of high flow. In the absence of salinity data, use specific conductance and temperature to calculate salinity.			
Secchi disk (meters)	00078	Report Secchi-depth transparency in meters to two significant figures (examples: 0.35 m or 1.3 m).			
Days since last significant precipitation (days)	72053	Report whole numbers. If it is raining when the sample is collected, or has rained within the last 24 hours, report a value of $< I$. Otherwise report the actual number, if known, or a 'greater than' value.			
E. coli (MPN/100 mL)	31699	First step: round the result to the nearest whole number to remove decimals. Second step: round to two significant figures (example: 347.1 to 347 to 350). Adjust < and > results based on dilution. Do not report "zero" but < 1.			
Enterococci (MPN/100 mL)	31701	First step: round the result to the nearest whole number to remove decimals Second step: round to two significant figures (example: 347.1 to 347 to 35(9.7 to 10). Adjust < and > results based on dilution. Do not report "zero" but < 1.			
Fecal coliform (colonies/100 mL)	31616	Always report fecal coliform densities as a whole number. If no colonies ard detected, report a less than value based on the volume filtered. For example report < 1 (100 mL filtered), < 4 (25 mL filtered), or < 10 (10 mL filtered). For concentrations > 100, report two significant figures. Do not record "TNTC" or "0/100 mL."			
Flow (Stream Discharge) (ft ³ /s)	00061	Report instantaneous flow values less than 10 but greater than 0.1 ft ² /s to the nearest tenth (example: 9.35 to 9.4). Report flow values greater than 1 ft ² /s to the nearest whole number (example: 20.62 to 2/). Actual flow value less than 0.1 ft ² /s but greater than or equal to 0.01 ft ² /s are reported and not subject to rounding (example: report 0.07 as 0.07). Report flow values < 0.01 ft ² /s as < 0.01. When there is no flow (pools), report 00061 as 0.0. When there is no water, do no treport a value for 000061.			
Flow severity (1—No Flow, 2—Low, 3—Normal, 4—Flood, 5—High, 6—Dry)	01351	Report flow severity for freshwater streams and rivers only. When there is no flow (pools), report a flow severity of I , and the instantaneous flow (00061) as $0.0 f \ell^3 k$. If the stream is dry, record only the flow-severity value of δ .			



Basic Parameters

 Observation 	ons
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- Field Parameters
- Conventional and Bacterial Parameters

ation ID:	Station Des	cription:						
llector(s) Nan	ie/Signature:							
te Collected:		Time Collect	ed;		Sample D	epth (meters	s):
Fi	eld Tests and Me	asurements:			Sample Ide	_		n:
	Water Temperature °C		00010	TAG ID			Sample ID	
	Specific Conductar	00094	Parameters Collected:			d:		
	pH (standard units	00400	x	E. Coli	X T. Phosphorus		hosphorus	
	Dissolved Oxygen	Dissolved Oxygen (mg/L)		x	TSS	x	Chlorophyll-a	
	Secchi Depth (met	ers)	00078	x	Ammonia-N	x	Pheo	ophytin-a
	Total Water Depth	(meters)	82903	x	Nitrate-N	x	Chloride	
	Instantaneous Str	eam Flow (cfs)	00061	x	Nitrite-N	x	Sulfate	
		Fiel	d Observ	atio	15:	-		
	01351 - Flow Seve	rity (1-no flow, 2- low						
	89835 - Flow mea	surement method	1-gage, 2-el	ectric,	3-mechanical, 4-weir/flume	:, 5-da	ppler)	
	72053 - Days since	e last significant rai	nfall					
	89966 - Present W	eather (1-clear, 2-p	artly cloudy,	3-clau	dy, 4-rain, 5-other)			
	89979 - Evidence	of Primary Contact	Recreation	(1 = 0	bserved, 0 = Not Observed			
	89978 - Primary C	ontact, Observed A	ctivity (Nur	nber d	F people abserved. Reporte	das 0-	10 or >	10)
	If sampling from a	Reservoir						
	00052 - Reservoir Stage (Feet Above Mean Sea Leve					rom TV	VDB we	bsite)
	0005	3 - Reservoir Perce	ent Full (col	lected	from TWDB website)			
	0005	1 - Reservoir Acce	ss Not Poss	ible, L	evel Too Low (Enter "1" I	ftrue)		
	If sampling from a	n perennial pool (is	solated poo	0				
	8986	i4 - Maximum pool	width in m	eters				
	89865 - Maximum pool depth in meters							
	89869 - Pool length in meters							
	8987	0 - Percentage the	pool cover	s with	in a 500 meter reach			
	74069 - Stream Flo	ow Estimate (cfs)			(W ×	D×L	×C÷T	= Flow Estimate
							(fee	
	Average Depth of Stream (D)							(fee
	Dista	nce Object Travels	(L)					(fee
	Corr	ection Factor (C)	(0.9 for	smoo	th or muddy bottom) (0	.8 for	rough	or rocky bottom
	Time	for Object to Trave	el Distance	(T)				(seconds
	Comments/Obser	vations:						



Observations

- Weather
- Water appearance/odor
- Wildlife and human activities
- Algae/macrophytes
- Etc.





Water Quality Monitoring in the Upper Neches Basin

- Currently, ANRA monitors 40 sites quarterly for field parameters, conventional parameters, and bacteria.
- Additional monitoring in the Upper Neches Basin is performed by TCEQ (Region 5 in Tyler and Region 10 in Beaumont) and the Lower Neches Valley Authority (LNVA).

Parameters for Quarterly Monitoring						
Field Parameters	Conventional Parameters	Bacterial Parameters				
Dissolved Oxygen	Ammonia-N	E. coli				
Days Since Last Significant Rainfall	Nitrate-N					
Flow Severity	Nitrite-N					
Instantaneous Stream Flow	Chloride					
Evidence of Primary Contact Recreation	Sulfate					
рН	Chlorophyll-a					
Present Weather	Pheophytin- <i>a</i>					
Secchi Transparency	Total Phosphorus					
Specific Conductance	Total Suspended Solids (TSS)					
Total Water Depth						
Water Temperature						



Water Quality Monitoring: Field Parameters

Field parameters are performed on-site by direct monitoring in the water body. This data is collected using multiprobe instrumentation or with a Doppler flow meter, for in-stream flow measurements.

- <u>Dissolved Oxygen (DO)</u> DO is a measure of the amount of dissolved oxygen that is available in the water. DO is vital for aquatic organisms to live. Where DO is too low, aquatic organisms may have insufficient oxygen to live.
- <u>Days Since Last Significant Rainfall</u> The days since the last significant precipitation can help explain anomalous data due to runoff. Significant precipitation would be any amount that would produce runoff and influence water quality. (http://water.weather.gov/precip/)
- <u>Flow Severity/Instantaneous Stream Flow</u> Flow is a measurement of the velocity of the water, measured in cubic feet per second (CFS). Flow combined with other parameters can be a good indicator of water quality. Flow can be affected by both natural and man-made sources.



Water Quality Monitoring: Field Parameters

- <u>Evidence of Primary Contact Recreation</u>—Are observations related to primary contact recreation that are needed to help the Water Quality Standards Group identify freshwater streams or rivers as suitable candidates for a Recreational Use Attainability Analysis. This information will also be considered as part of an existing RUAA to quantify the extent of existing primary contact recreation use.
 (https://www.tceq.texas.gov/assets/public/waterquality/swqm/monitor/manual/vol1/Chapt 3 revFY15.pdf)
- <u>pH</u> pH is a measure of the acidity or basicity of an aqueous solution. Most aquatic organisms are adapted to live within a specific pH range. pH can also affect the toxicity of many substances, which generally increase in solubility as pH decreases. The ability of water to resist changes in pH (its buffering capacity) is essential to aquatic life.
- <u>Present Weather</u> Clear, partly cloudy, cloudy, rain, or other. Documenting any recent meteorological events that may have affected water quality include heavy rains, a cold front, or very dry or very wet conditions.



Water Quality Monitoring: Field Parameters

- <u>Secchi Transparency</u> Secchi transparency remains an important secondary parameter for assessing eutrophication—the natural aging process in reservoirs and lakes—and for determining trends in water clarity. Eutrophication is accelerated by human activities that add nutrients to lakes, reservoirs, and the surrounding watersheds.
- <u>Specific Conductance</u> Specific Conductance is the measure of the water's capacity to carry an
 electrical current and is indicative of the amounts of dissolved solids present in a water body.
 Dissolved salt-forming substances such as sulfate, chloride, and sodium increase the conductivity of
 the water.
- <u>Total Water Depth</u> Measuring the depth of the water helps characterize a site. A site can be shallow, deep, or within a navigational channel. Rainfall and flow affect total water depth, so the total depth of a site can change depending on when it is sampled.
- <u>Water Temperature</u> Water temperature affects the oxygen content of the water (dissolved oxygen). Temperature also has an impact on cold-blooded animals. Water temperature may be affected by alterations to the riparian zone, changes in ambient temperature, and discharges.



Conventional parameters are also evaluated as part of the monitoring plan. During routine monitoring events, water samples are collected for laboratory analysis of conventional parameters. Conventional parameters include nutrients, minerals, and particulates. For routine monitoring stations, ANRA collects and analyzes samples for the following conventional parameters:

- <u>Ammonia-Nitrogen</u> (NH₃) Ammonia, which is produced from the breakdown of nitrogencontaining compounds, is found naturally in waters. In excess, algal blooms may occur. Elevated ammonia levels are indicative of organic pollution. These elevated levels can cause stress on aquatic organisms, as well as damage to tissue and gills.
- <u>Nitrate + Nitrite-Nitrogen</u> Elevated levels of nitrite and nitrate can produce nitrite toxicity in fish ("brown blood disease") and methemoglobinemia ("blue baby syndrome") in infants by reducing the oxygen-carrying capacity of blood. In surface water, high levels of nitrates can lead to excessive growth of aquatic plants. High levels of nitrates are also indicative of human-caused pollution.



The Nitrogen Cycle

TN=TKN + NO3- + NO2-Sources: Fertilizers, WWTP, Sewage





 <u>Chloride</u> - Chloride is one of the major inorganic ions in water and wastewater. It is an essential element for maintaining normal physiological functions in all organisms. Elevated chloride concentrations can adversely affect survival, growth, and/or reproduction of aquatic organisms.

• <u>Sulfate</u> - Sulfate is essential for plant growth, and low levels (under 0.5 mg/L) can be detrimental to algal growth. Excessive levels of sulfate can form strong acids and change the pH of the water. Excessively high levels may be toxic to cattle and other animals. Sulfate can also affect drinking water quality.



Nutrients - Chloride, Nitrite, Nitrate, Sulfate by Ion Chromatography (IC)

Theory of IC-What is chromatography? chroma (*Greek - colour*) graphein (*write*)

- Historically chromatography was a separation of a color in its elements.
- Today chromatography stands for a wide range of physicochemical separation processes in which elements of a sample are separated, detected, identified and quantified.



Anions = lons having a negative charge



Nutrients (Chloride, Nitrite, Nitrate, Sulfate) by Ion Chromatography

Ingredients for IC

Eluent, Pump, Sample Injection, Separation, Detector

- The mobile phase (eluent) is pumped through the system with a constant flow
- The sample is injected into it
- The mobile phase carries the sample through the static phase (separator) where the sample is split up into its component ions
- In the detector single components are recognized by a change in conductivity

Physical Chemistry of Chromatography

• components are separated by interaction with a stationary and a mobile phase





Nutrients (Chloride, Nitrite, Nitrate, Sulfate) by Ion Chromatography

I Column (separation column) The separation of a sample takes place in the stationary phase inside the column. As it passes through the stationary phase, each ion will be retained for a certain amount of time before passing through to the detector. This time is called **retention time**. For proper identification and quantification, each ion should have a unique retention time .

Columns are accessories and can not be serviced. Sometimes they can be regenerated according to the manufacturers instructions.

i-Columns are an important part of intelligent IC technology, using an integral column chip to provide:

- traceability of the column
- monitoring options
- warning messages to protect the column
- availability of column properties
- customer tailored column handling







Nutrients (Chloride, Nitrite, Nitrate, Sulfate) by Ion Chromatography

Features of the chromatogram:

- The **injection dip** is a drop in conductivity caused by the water carrying sample passing through the detector as the sample ions are retained by the column. It is an indicator that a sample injection has been made.
- Each ion has a different **relative conductivity**; this means some ions will have a much higher response than others at the same concentration (as shown below).





 <u>Chlorophyll-a/Pheophytin-a</u> - Chlorophyll-a is an indicator of algal biomass in a water body. Increased concentrations indicate potential eutrophication or nutrient loading. Diurnal shifts in DO and pH resulting from increased photosynthesis and respiration can cause stress to aquatic organisms.





• <u>Total Phosphorous (TP)</u> -Phosphorus is essential to the growth of organisms, and is considered a growth-limiting nutrient. Elevated levels in water may stimulate the growth of photosynthetic aquatic macro and microorganisms. Elevated phosphorus levels contribute to eutrophication and may cause algal blooms.



ANGELINA & NECHES RIVER AUTHORITY



 <u>Total Suspended Solids (TSS)</u> – TSS is a measure of the total suspended particles in water (Non-Filterable Solids). High levels of TSS increase the turbidity of the water, reducing light penetration which subsequently decreases oxygen production by plants.





Water Quality Monitoring: Bacterial Parameter

<u>Escherichia coli (E. coli)</u> - E. coli is an indicator of fecal contamination. Fecal contamination is a health concern to the general public, and its presence indicates a risk for contact recreation. The presence of *E. coli* in the water indicates that pathogenic organisms may be present.

Overview of the analysis steps for enumeration of *E. coli* bacteria using Colilert® media (IDEXX)



1. Add Colilert to sample



2. Pour sample/Colilert Mixture into quanti-tray



3. Seal tray and incubate 24 hours



4. Count positive wells and refer to MPN table



Water Quality Monitoring: It's all Related





Questions?

Hannah Lucia Quality Assurance Officer/Technical Director Angelina & Neches River Authority <u>hlucia@anra.org</u> 936-633-0063

References and Resources:

- Surface Water Quality Monitoring Procedures, Volume 1: Physical and Chemical Monitoring Methods(https://www.tceq.texas.gov/assets/public/comm_exec/pubs/rg/rg415/rg-415.pdf)

 Water Quality: What We Collect and Why, A. Kilpatrick, Trinity River Authority (https://www.tceq.texas.gov/assets/public/waterquality/swqm/monitor/training/swqmworkshop 30/Wed_Kilpatrick.pdf)

- Metrohm Training and Installation-850/881/882 IC- Basic IC Theory and Flow Path